GLUE THIS PAGE TO YOUR NOTEBOOK





- LESS 02

SEDIMENT

BETTER FOR

AQUATIC LIFE!

- MORE 02 - LESS SEDIMENT

- MORE

WATER SYSTEM HEALTH

When it comes to the health of a water system, there are many factors that go into it. The balance between physical, chemical and biological variables determines the health of a water system.

PHYSICAL	CHEMICAL	BIOLOGICAL	These variables
 Temperature Turbidity Water Movement 	 Dissolved oxygen (+ other gases) pH Nitrates Salinity 	FishAlgaeInsectsPlants	are subject to change from both natural and man-made forces

- Freshwater is a major concern because it is the main source of water for humans and animals
- Our freshwater that we use can be <u>SAFE</u> or <u>POTABLE</u> O



PHYSICAL INDICATORS

TEMPERATURE

- The temperature of a body of water determines the organisms that can live there
- Many organisms have a preferred temperature range where they will thrive
 - THINK ABOUT IT! We cannot thrive in an environment that is -50°, but some organisms can!
- As the temperature of the water increases:
 - o It is able to dissolve more sediment which can block the light and not allow photosynthesis to occur
 - o It dissolves LESS oxygen (because particles are moving too fast and O_2 can escape into the air) and may not contain enough for organisms to survive

TURBIDITY

Turbidity is how clear/cloudy a body of water is.

- Cloudiness is due to the amount of sediment dissolved in the water
- A high turbidity = not potable
 - o Can lead to increased temperatures, decreased DO, and impairment of some aquatic organisms



CHEMICAL INDICATORS

DISSOLVED OXYGEN

Dissolved oxygen is the amount of oxygen in water that is available for aquatic organisms to use.

- The level of oxygen in surface water is important for many organisms such as zooplankton and fish to thrive.
- There are two ways oxygen gets into the water:
 - 1. From the AIR (being trapped by waves and moving currents)
 - 2. From PLANTS releasing O2 during photosynthesis

<u>pH LEVEL</u>

The pH of a body of water determines how acidic or how basic it is.

- pH is measured on a scale from 0-14 ACIDIC ALKALINE
 - \circ 0-6 = ACIDIC
 - \circ 7 = NEUTRAL
 - \circ 8-14 = BASIC (Alkaline)
 - The pH of water is known to have a synergistic effect, which means that materials (iron, aluminum, ammonia, mercury) introduced into bodies of water can have more or less of an impact based on the pH of the water.
 - o <u>EXAMPLE</u>: Metals in more acidic water can become more dangerous and more poisonous than they normally would be in neutral water.

NITRATES + PHOSPHATES

Nitrates and phosphates come from Nitrogen and Phosphorous, which are essential nutrients for healthy plant growth.

- Too many nitrates or phosphates in drinking water can make it unhealthy

SOURCES OF NITRATES	SOURCES OF PHOSPHATES	
 Runoff contaminated with fertilizers Septic tank leaks Sewage Natural deposit erosion 	 Human and animal waste Laundry Cleaning and industrial waste 	

SALINITY

Salinity is the measure of salt in water and can be an indicator of how healthy a water system is.

- Salinity can enter water systems through natural processes of weathering rocks from wind and rain
- High concentrations of salinity can cause vegetation to become unhealthy or die and can lead to a decrease in biodiversity .



、 AMOUNT OF DIFFERENT SPECIES LIVING IN AN AREA





BIOLOGICAL INDICATORS

Biological indicators (Bioindicators) are macroinvertebrates that can give an indication of how healthy a water system is.

- The presence and numbers of the types of fish, insects, algae, plants and other aquatic organisms can tell us how healthy the water they live in is
- These organisms are usually easy to collect and identify
- These organisms are used to measure water health because many are very sensitive to pollution
 - Poor water quality is indicated by a few number of bioindicator organisms in one place





The species in the GREEN to the right is a bioindicator species because it is disturbance-intolerant (which means it does not handle a change in the water well).

- As the water withdrawal occurs, and the water system is less healthy for these organisms, their abundance numbers begin to drop.

WATER STEWARDSHIP

There has only been a growing awareness and concern for water pollution for the past 45-50 years. Before that, there was little concern about what was being put or dumped into our water systems.

- Part of this awareness came with the development of the Environmental Protection Agency (EPA) in 1970
- In 1972, the Clean Water Act established the regulations on putting pollutants into the water
 - This gave the EPA the authority to test for pollutants and chemicals in the water and set maximum amounts allowed to be found in the water



the water

actually set fire!

BEFORE THE EPA



AFTER THE EPA



Now you can swim in that same river!

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8.E.1.3 – WATER QUALITY guided notes



- MORE

SEDIMENT

BETTER FOR AQUATIC

LIFE!

- MORE 02 - LESS SEDIMENT

WATER SYSTEM HEALTH

When it comes to the _____ of a water system, there are many factors that go into it. The balance between _____, chemical and _____ variables determine the health of a water system.

1			K)
PHYSICAL	CHEMICAL	BIOLOGICAL	These variables
 Temperature Water Movement 	 Dissolved oxygen (+ other gases) pH Salinity 	FishAlgaeInsectsPlants	are subject to from both natural and forces

- Freshwater is a major _____ because it is the source of water for humans and animals
- Our freshwater that we use can be _____ or _____





PHYSICAL INDICATORS

- The _____ of a body of water determines the _____ that
- Many organisms have a _____ temperature range where they will thrive
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TURBIDITY

- _____ is how clear/____ a body of water is.
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CHEMICAL INDICATORS

DISSOLVED OXYGEN

Dissolved _____ is the amount of oxygen in water that is _____ for aquatic organisms to use.

- The level of oxygen in surface water is important for many organisms such as _____ and fish to thrive.
- There are _____ ways oxygen gets into the water:
 - 3. From the ____ (being trapped by waves and moving currents)
 - 4. From _____ releasing O_2 during

<u>ph level</u>

The ____ of a body of water determines how _____ or how _____ it is. ALKALINE

- pH is measured on a scale from 0-14 ______
 - o 0-6 = ____
 - o 7 = ____
 - $0 \ 8-14 = ____$ (Alkaline)

- The pH of water is known to have a _____ effect, which means that materials (iron, aluminum, ammonia, mercury) introduced into bodies of water can have _____ or _____ of an impact based on the pH of the water.

NEUTRAL

o EXAMPLE: _____ in more acidic water can become more dangerous and more _____ than they normally would be in neutral water.

NITRATES + PHOSPHATES

_____ and _____ come from Nitrogen and Phosphorous, which are _____ nutrients for healthy plant growth.

- Too many nitrates or phosphates in drinking water can make it _____

SOURCES OF NITRATES	SOURCES OF PHOSPHATES	
 Runoff contaminated with Septic tank leaks Natural deposit erosion 	 Human and animal Cleaning and industrial waste 	

SALTNTTY

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- Salinity can enter water systems through natural processes of _____ rocks from wind and rain
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AMOUNT OF DIFFERENT

AREA

LIVING IN AN

BIOLOGICAL INDICATORS

Biological indicators (_____) are _____) are ______ that can give an indication of how healthy a water system is.

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The species in the ______ to the right is a bioindicator species because it is ______ (which means it does not handle a change in the water well). - As the water withdrawal occurs, and the water system is _____ healthy for these organisms, their ______ numbers begin to

WATER STEWARDSHIP

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- Part of this awareness came with the development of the _____
 - _____ (EPA) in 1970
- In 1972, the _____ established the regulations on putting _____ into the water
 - This gave the EPA the authority to test for pollutants and chemicals in the water and set _____ amounts allowed to be found in the water



was so _____ with oils and chemicals that the water actually set ____!





AFTER THE EPA



Now you can swim in that same river!